

CHEMISTRY 212 - LABORATORY SCHEDULE
Fall 2017

TEXT/MATERIALS: Laboratory worksheets and POGIL handouts are available on the Blackboard website. Lab manual: *CHEM 212*, Fountainhead Press (2017); ISBN 978-1-68036-497-2.

REQUIRED EYE PROTECTION: Students must wear the approved goggles issued by the department *at all times in the laboratory—NO EXCEPTIONS.*

Week of:	EXPERIMENT
1. Aug. 28 th	CHECK-IN / SAFETY ORIENTATION : Safety in the Laboratory; Glassware & Equipment/Data Representation and Recording
2. Sept. 4 th	LABOR DAY (Holiday – no lab meetings for the week.)
3. Sept. 11 th	Separation and Recovery of Components in a Ternary Mixture
4. Sept. 18 th	The Bohr Model and Spectroscopy of the Hydrogen Atom
5. Sept. 25 th	Periodic Properties of Some Elements
6. Oct. 2 nd	Lewis Structures and Molecular Shape
7. Oct. 9 rd	Determining the Empirical Formula of Magnesium Oxide
8. Oct. 16 th	LAB MIDTERM EXAM
9. Oct. 23 rd	Stoichiometry - A Mole Ratio Study
10. Oct. 30 th	Classifying Chemical Reactions
11. Nov. 6 th	Redox Chemistry - Activity of Metals
12. Nov. 13 th	Calorimetry and Hess's Law
13. Nov. 20 th	THANKSGIVING BREAK
14. Nov. 27 th	Determination of the Universal Gas Constant, R / CHECK OUT
15. Dec. 4 th	LAB FINAL EXAM

***FAILURE TO CHECK OUT MAY RESULT IN A FAILING GRADE FOR THE ENTIRE SEMESTER.**

Grading: The overall lab grade is a weighted average, and is calculated using the formula below:

$$(\text{Lab average} \times 0.70) + (\text{Midterm Exam} \times 0.15) + (\text{Final Exam} \times 0.15) = \text{lab grade}$$

Letter grades are assigned based on the overall lab grade compared to the following cutoffs:

$$90\% = \text{A}; \quad 80\% = \text{B}; \quad 70\% = \text{C}; \quad 60\% = \text{D}; \quad <60\% = \text{F}$$